Name:	ass:	Date:

H.W. - Unit 2 Study Guide

This Study Guide is due Tuesday, November 3rd. You will need to visit <u>www.MrAscience.com</u> to answer most of the questions!

Topics:

H.W. Finish Study Guide

H.W. None

		October 5:	October 6:	October 7:	October 8:	October 9:
					H.W. Finish Take 5s	H.W. None
6.	Pla	ying Doctor				
5.	5. Elements, Compounds, & Mixtures					
4.	Ionic and Covalent Bonding			12000000	mitro to onomisti	
3.	Lewis Dot Structures			Take 5s	Intro to Chemistry	
2.	Pro	tons, Neutrons, &	Electrons			
1.	My	stery Mix			October 1:	October 2:

			H.W. Finish Take 5s	H.W. None
October 5:	October 6:	October 7:	October 8:	October 9:
Lab Procedures & Chemical I.D.	Mystery Mix Day #1	Mystery Mix Day #2	Mystery Mix Day #3	Mystery Mix Day #4
H.W. Write Paragraph	H.W. None	H.W. Myst. Mix Elements	H.W. None	H.W. Mini-Essay
October 12:	October 13:	October 14:	October 15:	October 16:
Columbus Day	Protons and Electrons	Neutrons	Introduce Atom Modeling	Atom Modeling Competition (Mr. A absent)
	H.W. None	H.W. Element Math	H.W. None	H.W. None
October 19:	October 20:	October 21:	October 22:	October 23:
Lewis Dot Diagrams	Ionic Bonding	Covalent Bonding	The Element Dating Game	Elements, Compounds, & Mixtures
H.W. None	H.W. None	H.W. Real-World Bonding	H.W. None	H.W. Checkpoint Quiz
October 26:	October 27:	October 28:	October 29:	October 30:
Playing Doctor: Introduction	Playing Doctor: Simulation	Review Carousel (Interrupted Day)	Playing Doctor: Begin Reports	Playing Doctor: Finish Reports
H.W. None	H.W. None	H.W. None	H.W. None	H.W. Finish Reports
November 5:	November 3:			
Review Day	Unit 2 Test			

Day-by-Day Recap:

Go to www.MrAscience.com and visit the entry for each day. After reading it, answer the questions to the right.

October 2: Intro to Chemistry	Of the three examples we discussed, which one had to do with (i)combining atoms? (ii)splitting atoms? (iii)discovering new compounds?	
October 5: Lab Procedures & Chemical I.D.	What were the four methods of Chemical I.D. mentioned in the article (Page 203)?	
October 6: Mystery Mix Day 1	Of those four methods, which did you use on Day 1 of the Mystery Mix Lab?	
October 7: Mystery Mix Day 2	Choose one of our nine white substances. Give its common name, chemical name, and chemical formula.	
October 8: Mystery Mix Day 3	Which combination of chemicals did Thomas and Jake test in the video? What was the result of their test?	
October 9: Mystery Mix Day 4	Other than the Mystery Mixture, all of the nine white substances were (elements, compounds, or mixtures)	
October 13: Protons and Electrons	How many electrons can be held (at most) in each of the first four electron shells?	

October 14: Neutrons	"To figure out the number of neutrons in any normal atom, you"
October 15: Intro to Atom Modeling	Draw a complete model of a Sodium atom, below. Include protons, neutrons, and electrons.
October 16: Atom Modeling Competition	
October 19: Lewis Dot Diagrams	Draw the Lewis Dot Structure of a Sodium atom, below. #somucheasier
October 20: Ionic Bonding	On Molly's worksheet, what were the only two elements that formed an ionic bond?
October 21: Covalent Bonding	Summarize the story of Henry and Fiona in scientific terms.
October 22: Element Dating Game	What two elements bonded in the video? What type of bond did they form?

October 23: Elements, Compounds, and Mixtures	Give an example of an element, a compound, and a mixture from the picture on the website.			
October 26: Playing Doctor: Introduction	How much has human life expectancy increased since 1840?			
October 27: Playing Doctor: Simulation	In the video, what was one mistake that a student made?			
October 28: Review Carousel	Go back to Review Carousel Question #6, where you drew a molecule of ammonia. How did you know that ammonia would bond covalently?			
October 29: Playing Doctor: Begin Reports	Write the o	correct prescriptions Henry Allen	s for the eight pation	ents below. Riley Moeykens
October 30: Playing Doctor: Finish Reports	Helen Shapiro	Sarah Clarkson	John Ferro	Lucas Morrison
<u>November 2:</u> Review Day	Go back to the front page of the study guide. Which topic has given you the most trouble during Unit 2? Now go back through the website and study it extra!			