

Name: _____

Class: _____

Date: _____

H.W. - Unit 1 Study Guide

This Study guide is due tomorrow, Sept. 30th. You will need to visit www.MrAscience.com to answer most of the questions!

Topics:

1. The Meter
2. Converting Metric Units

3. Volume
4. Mass

5. Density
6. Boats

Class Calendar:

	<u>September 1:</u> Introduction to 8 th Grade Science H.W. None	<u>September 2:</u> DDM Pre-Tests H.W. Get 3-Ring Binder	<u>September 3:</u> The Meter H.W. None	
	<u>September 8:</u> Converting Metric Units H.W. None	<u>September 9:</u> Volume H.W. None	<u>September 10:</u> Mass H.W. Website Check	<u>September 11:</u> The Mass of Water Lab H.W. None
<u>September 14:</u> Density Lecture (Picture Day) H.W. Checkpoint Quiz	<u>September 15:</u> Density Lab H.W. None	<u>September 16:</u> Changes in Density H.W. Finish Post-Lab	<u>September 17:</u> Go Over H.W. H.W. None	<u>September 18:</u> Review Carousel H.W. None
<u>September 21:</u> Intro to Boats H.W. None	<u>September 22:</u> Building Boats H.W. None	<u>September 23:</u> Boat Testing H.W. Begin Boats Essay	<u>September 24:</u> How Much Will It Hold? Day #1 H.W. Boats Essay	<u>September 25:</u> The Finest Hours Day H.W. Finish Boats Essay
<u>September 28:</u> How Much Will It Hold? Day #2 H.W. None	<u>September 29:</u> Review Day H.W. Study Guide	<u>September 30:</u> Unit 1 Test (Mr. A absent) H.W. None		

Day-by-Day Recap:

Go to www.MrAscience.com and visit the journal entry for each day.

After reading it, answer the “key question” to the right...

<u>September 1:</u> Intro to Grade 8 Science	What is the name of the device in the picture? windmill Which of this year’s seven units is it related to? Environmental Science
<u>September 2:</u> DDM Pre-Tests	Why did Mr. A give all three DDM pre-tests at once? To same time (and still find an average score for all three).
<u>September 3:</u> The Meter	“Part of being a good scientist is...” *Hint: It’s a quote from that day’s entry...
<u>September 8:</u> Converting Metric Units	Fill in the 6 blanks below (3 prefixes and 3 basic units). <p>Ladder Method</p> <p>To convert to a smaller unit, move decimal point to the right or multiply.</p> <p>To convert to a larger unit, move decimal point to the left or divide.</p> <p>See metric ladder (Page 103).</p>
<u>September 9:</u> Volume	What is volume? See website. What units do we use to measure it in (a) the English system and (b) the Metric System? (a) Gallons and ounces (b) Liters, milliliters, etc.

<p><u>September 10:</u> Mass</p>	<p>What is mass? See website.</p> <p>What units do we use to measure it in (a) the English system and (b) the Metric System? (a) pounds (b) grams</p>
<p><u>September 11:</u> The Mass of Water Lab</p>	<p>Explain how the centimeter, the milliliter, and the gram are all related. See website.</p>
<p><u>September 14:</u> Density Lecture</p>	<p>In the video, how did they determine what percentage of the object would float above the water line? If the density is 0.5 g/ml, that means 50% of the object would be above water.</p>
<p><u>September 15:</u> Density Lab</p>	<p>What are the formula <u>and</u> the units for density? $D = M \div V$ (g/ml)</p> <p>How do we determine whether an object will float in water? If it is below 1.0 g/ml it will float.</p>
<p><u>September 16:</u> Changes in Density</p>	<p>In the video, Phoebe shows us three different-colored samples. Which ones floated and which ones sank? 1. Hot water (red) floated. 2. Cold water (blue) sank. 3. Salt water (purple) sank.</p>
<p><u>September 17:</u> Go Over H.W.</p>	<p>Click the link to view the “Post-Lab Answers.” What is one notable exception mentioned in Question #3? Most substances get more dense when they cool off. But ice is a notable exception; it expands because the molecules form ice crystals.</p>
<p><u>September 18:</u> Review Carousel</p>	<p>What were the three steps to figuring out the blimp problem in the Review Carousel? 1. Convert mass to kilograms. 2. Calculate density of blimp. 3. See if blimp was less dense than CO₂.</p>

<u>September 21:</u> Intro to Boats	<p>The aircraft carrier in the picture has a mass of 200,000,000 lbs. (or 90,718,474,000 grams). So we know that its volume must be <i>at least</i>...</p> <p>The volume must be at least 90,718,474,000 ml for it to float.</p>
<u>September 22:</u> Building Boats	<p>If you had to rebuild your boat again, what would you do differently?</p> <p>No help on this one!</p>
<u>September 23:</u> Boat Testing	<p>What were the three keys to boat-building that Mr. A discussed in the video?</p> <p>Watch the darn video, you lazy bum!</p>
<u>September 24:</u> How Much Will It Hold? – Day 1	<p>What is the modified formula we used in this lab?</p> $d = \frac{m_b + m_c}{v}$
<u>September 25:</u> The Finest Hours Day	<p>The plastic boat in the picture below has a mass of 100 grams. How many bolts could it hold before sinking?</p> <p>V = 360 ml</p>
<u>September 28:</u> How Much Will It Hold? – Day 2	<div data-bbox="824 1440 1143 1581" data-label="Image"> </div> $1.0 = \frac{100 + m_c}{360}$ <p>$m_c = 260 \text{ g or } 4 \text{ bolts}$</p>
<u>September 29:</u> Review Day	<p>What is the one topic that you need the most work on before tomorrow's test?</p> <p>It's up to you.</p>